Teacher: Nidya Caviedes

Period(s): 2,3,4

Unit Title: ATOM State Standards: H.C.1A.2

|           | Standards | Goals As a result of this lesson the student will be able to:   | Instructional Activities (aligned,<br>Plan sequenced, build, time)   | S<br>t<br>u<br>d<br>e<br>(Thinking & Problem<br>n<br>Solving, Real World)<br>t<br>W<br>o<br>r<br>k   | Asse<br>ssme (aligned, rubrics, >2,<br>nt written) | Grouping Method                         | Materials  | Accommodations<br>(IEP, 504, ESOL)  |
|-----------|-----------|---|--|--|--|---|--|---|
| Monday    | H.C.2A.1  | Obtain and communicate<br>information to describe and<br>compare subatomic particles with<br>regard to mass, location, charge,<br>electrical attractions and<br>repulsions, and impact on the<br>properties of an atom. | Prepared warm-up questions.<br>How to count atoms mini<br>lesson<br>Practice worksheet                                       | How do I know how<br>many atoms of each<br>element are in a<br>compound?<br>Practice counting the<br>number of atoms in<br>various models                                      | Individual practice<br>Whole group                 | Whole group<br>Assigned<br>small groups | Notebook<br>Textbook<br>Worksheet                          | Extended time<br>on assignments.<br>Read aloud all<br>directions from<br>handouts                 |
| Tuesday   | H.C.2A.1  | Obtain and communicate<br>information to describe and<br>compare subatomic particles with<br>regard to mass, location, charge,<br>electrical attractions and<br>repulsions, and impact on the<br>properties of an atom. | Prepared warm-up questions.<br>Counting atoms worksheet<br>review<br>Google Classroom<br>Atomic Structure tutorial           | Count the number of<br>atoms of an element in a<br>molecule and can<br>represent the molecules<br>on a visual with 80% or<br>higher accuracy.<br>Work by using Virtual<br>labs | Formal<br>Atomic Structures<br>Molar mass          | Whole group<br>Assigned<br>small groups | Notebook<br>Textbook<br>Worksheet<br>Computer<br>Materials | Extended time<br>on assignments.<br>Read aloud all<br>directions from<br>handouts.                |
| Wednesday | H.C.2A.1  | Obtain and communicate<br>information to describe and<br>compare subatomic particles with<br>regard to mass, location, charge,<br>electrical attractions and<br>repulsions, and impact on the<br>properties of an atom. | Prepared warm-up questions.<br>Gram and mole conversions<br>Mini lesson<br>Students take notes<br>Mole worksheet<br>Feedback | Practice Gram/mole<br>conversion   | Individual practice<br>Whole group<br>Molar mass   | Whole group<br>Assigned<br>small groups | Notebook<br>Textbook<br>Worksheet<br>Computer<br>Materials | Extra time will<br>be given as<br>needed, one to<br>one interactions<br>as needed or<br>requested |

| Thursday | H.C.2A.1 | Obtain and communicate<br>information to describe and<br>compare subatomic particles with<br>regard to mass, location, charge,<br>electrical attractions and<br>repulsions, and impact on the                           | Prepared warm-up questions.<br>Counting atoms review<br>Mole, Molar mass Review | Student work with a partner. | Individual practice<br>Whole group<br>Counting atoms<br>Molar mass | Whole group<br>Assigned<br>small groups | Textbook<br>Notebook<br>notes | Extra time will<br>be given as<br>needed, one to<br>one interactions<br>as needed or                           |
|----------|----------|---|---|------------------------------|--|---|-------------------------------|--|
| Fridav   | H.C.2A.1 | Obtain and communicate<br>information to describe and<br>compare subatomic particles with<br>regard to mass, location, charge,<br>electrical attractions and<br>repulsions, and impact on the<br>properties of an atom. | Prepared warm-up questions.<br>Quiz   | Quiz                         |  | Whole group<br>Assigned                 | Quiz                          | requested<br>Extra time will<br>be given as<br>needed, one to<br>one interactions<br>as needed or<br>requested |

\* All plans are subject to change. Student progress will be monitored and adjustments will be made.